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Atom. Smallest particale of
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electrons. Nucleus. The
small, positively charged
center of the atom.

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(Properties Of Atoms and The

The

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The Periodic Table Parts of
an Atom • An atom consists
of a nucleus surrounded by
one or more electrons •
Atoms are electrically

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neutral w/ the same number
of protons as electrons. •
The Majority of the atom is
empty space. If nucleus were
the size of a pencil eraser,
the closest electron would
be 100 yards away!

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Chapter 16 Properties Of Atoms And

BeccaSwan1. Physical Science
Chapter 16 - Properties of
Atoms and the Periodic
Table. atom. nucleus.
proton. neutron. Smallest
unit of matter that retains
the properties of an elem...

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A region that is located at the center of an atom and contains... A subatomic particle that has a positive charge and that is fo...

chapter 16 section 1
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490 CHAPTER 16 Solids,

Liquids, and Gases Liquid

Figure 4 The particles in a
liquid are moving more

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freely than the particles in a solid. They have enough kinetic energy to slip out of the ordered arrangement of a solid. Figure 5 In gases, the particles are far apart and the attractive forces between the particles

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are over-come. Gases do not
have a defi-

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the Periodic Table. atom.

nucleus. proton. neutron.

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retains the properties of an
elem... A region that is
located at the center of an

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atom and contains... A
subatomic particle that has
a positive charge and that
is fo...

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Section 1 Answers
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